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Acta Cryst. (1993). **C49**, 1761–1762

Structure of the Copper(II) Complex of Isonicotinic Acid

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(Received 25 September 1992; accepted 25 March 1993)

Abstract

In crystals of diaquabis(isonicotinato-*N*)copper(II) dihydrate, the Cu atom is four-coordinate with the two N atoms of the pyridine rings and the two O atoms of the two water molecules in a *trans* orientation. The coordination geometry is square planar with distances Cu—O 1.985 (2) and Cu—N 2.004 (2) Å.

Comment

Both isoniazid and iproniazid are well known isonicotinic acid derivatives which are used as anti-tuberculosis drugs (Carrington, Bird & Levence, 1984; Pinelopi, 1988), and have bacterial mutagenicity (Parodi *et al.*, 1981). These derivatives inhibit copper(II)-containing serum amine oxidase (Morpurgo *et al.*, 1988; Masuda, Nakamura & Shimomura, 1990). In order to obtain structural information on the mode of interaction between copper(II)-containing amine oxidase and isoniazid and/or iproniazid, we thought it worthwhile to investigate the crystal structure of the complex of isonicotinic acid with copper(II). To date, the crystal structures of the complexes of isonicotinic acid with calcium(II) (Cole & Holt, 1989) and with copper(I)

chloride (Goher & Mak, 1985) have been determined, but the complex with copper(II) has not been subjected to crystal-structure analysis.

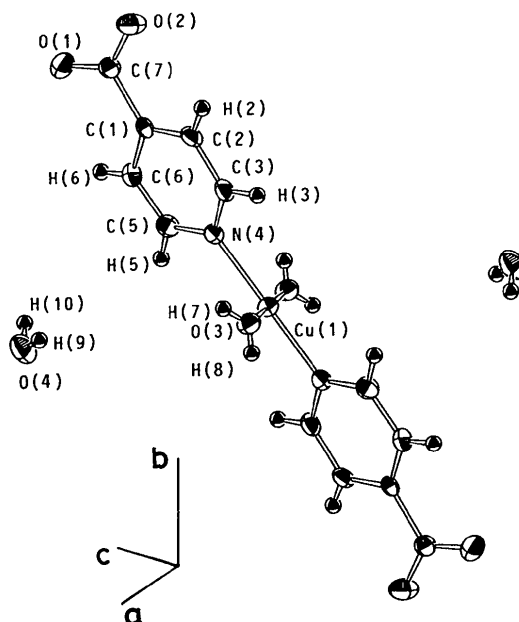


Fig. 1. Perspective view of diaquabis(isonicotinato)copper(II) dihydrate with atomic numbering, along the *a* and *c* axes.

Experimental

Crystal data

[Cu(C₆H₄NO₂)₂(H₂O)₂].-
2H₂O

M_r = 379.81

Triclinic

P $\bar{1}$

a = 6.895 (1) Å

b = 9.181 (1) Å

c = 6.3377 (8) Å

α = 105.24 (1)°

β = 108.20 (1)°

γ = 99.45 (1)°

V = 354.2 (1) Å³

Z = 1

D_x = 1.781 Mg m⁻³

D_m = 1.780 (1) Mg m⁻³

Data collection

Rigaku AFC-5R diffractometer
 ω -2 θ scans

Absorption correction:

DIFABS (Walker & Stuart, 1983)

T_{min} = 0.79, *T_{max}* = 1.29

1756 measured reflections

1621 independent reflections

1487 observed reflections

[*I* > 3 σ (*I*)]

Mo *K* α radiation

λ = 0.71069 Å

Cell parameters from 25
reflections

θ = 43.5–48.3°

μ = 1.588 mm⁻¹

T = 296 K

Plate

0.40 × 0.40 × 0.30 mm

Blue

Crystal source: solution of
isonicotinic acid–CuCl₂
(1:7) in 10% ethanol

R_{int} = 0.013

θ_{\max} = 55.0°

h = 0 → 8

k = -11 → 11

l = -8 → 7

3 standard reflections

monitored every 150
reflections

intensity variation: none

Refinement

Refinement on F^2 Final $R = 0.033$ $wR = 0.041$ $S = 1.68$

1487 reflections

138 parameters

All H-atom parameters refined

$$w = 4F_o^2/\sigma^2(F_o^2)$$

$$(\Delta/\sigma)_{\max} = 0.05$$

$$\Delta\rho_{\max} = 0.60 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.69 \text{ e } \text{\AA}^{-3}$$

Atomic scattering factors from *International Tables for X-ray Crystallography* (1974, Vol. IV)Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (\AA^2)
$$U_{eq} = \frac{1}{3} \sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$$

	x	y	z	B_{eq}
Cu(1)	0.0	0.0	0.0	1.68 (1)
O(1)	0.4383 (3)	0.7143 (2)	0.9298 (3)	2.86 (6)
O(2)	0.2980 (4)	0.7944 (2)	0.6267 (3)	2.87 (6)
O(3)	0.2043 (3)	0.0413 (2)	-0.1530 (3)	2.04 (5)
O(4)	0.7320 (4)	0.0943 (3)	0.7552 (4)	3.21 (7)
N(4)	0.1253 (3)	0.2183 (2)	0.2323 (3)	1.67 (5)
C(1)	0.2719 (4)	0.5246 (3)	0.5465 (4)	1.59 (6)
C(2)	0.1974 (4)	0.4940 (3)	0.3050 (4)	1.91 (7)
C(3)	0.1277 (4)	0.3412 (3)	0.1545 (4)	1.92 (7)
C(5)	0.2012 (4)	0.2483 (3)	0.4664 (4)	2.10 (7)
C(6)	0.2769 (4)	0.3980 (3)	0.6266 (4)	1.96 (7)
C(7)	0.3431 (4)	0.6915 (3)	0.7157 (4)	1.94 (7)

Table 2. Geometric parameters (\AA , $^\circ$)

Cu(1)—O(3)	1.985 (2)	C(1)—C(2)	1.386 (3)
Cu(1)—N(4)	2.004 (2)	C(1)—C(6)	1.386 (3)
O(1)—C(7)	1.251 (3)	C(1)—C(7)	1.516 (3)
O(2)—C(7)	1.255 (3)	C(2)—C(3)	1.377 (3)
N(4)—C(3)	1.345 (3)	C(5)—C(6)	1.376 (3)
N(4)—C(5)	1.343 (3)		
O(3)—Cu(1)—O(3)	180.00	C(6)—C(1)—C(7)	121.6 (2)
O(3)—Cu(1)—N(4)	89.87 (8)	C(1)—C(2)—C(3)	119.6 (2)
N(4)—Cu(1)—N(4)	180.00	N(4)—C(3)—C(2)	122.5 (2)
Cu(1)—N(4)—C(3)	119.6 (2)	N(4)—C(5)—C(6)	122.7 (2)
Cu(1)—N(4)—C(5)	122.6 (2)	C(1)—C(6)—C(5)	119.5 (2)
C(3)—N(4)—C(5)	117.8 (2)	O(1)—C(7)—O(2)	125.9 (2)
C(2)—C(1)—C(6)	117.8 (2)	O(1)—C(7)—C(1)	117.7 (2)
C(2)—C(1)—C(7)	120.6 (2)	O(2)—C(7)—C(1)	116.4 (2)

Data collection, cell refinement: *Rigaku MSC/AFC Data Collection and Refinement Software* (Rigaku Corporation, 1988). The scan rate was $32^\circ \text{ min}^{-1}$ in ω and the scan width was $(1.68 + 0.30 \tan \theta)^\circ$. The ratio of peak counting time to background counting time was 2:1. Programs used to solve structure: *MULTAN87* (Debaerdemaeker, Germain, Main, Tate & Woolfson, 1987) and *DIRDIF* (Beuskens, 1984). All calculations including data reduction: *TEXSAN* (Molecular Structure Corporation, 1985). Refinement was by full-matrix least squares.

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates, complete geometry, bond distances and angles involving H atoms, and torsion angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71216 (18 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: OH1018]

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Acta Cryst. (1993). **C49**, 1762–1764

Structure of $[\text{Ru}(\eta^6\text{-C}_7\text{H}_8)(\text{acetylacetonato})\text{Cl}]$

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(Received 26 August 1992; accepted 22 March 1993)

Abstract

The structure of (acetylacetonato-*O,O'*)chloro(η^6 -cyclohepta-1,3,5-triene)ruthenium(II), $[\text{Ru}(\eta^6\text{-C}_7\text{H}_8)\{\text{HC}(\text{COMe})_2\}_2\text{Cl}]$, was determined. The Ru atom lies on a crystallographic mirror plane which is coincident with the molecular mirror plane. The geometry at the Ru atom is, as expected for ruthenium(II), approximately octahedral, with O—Ru—Cl and O—Ru—O bond angles of $84.1(1)^\circ$ and $90.4(1)^\circ$, respectively.

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